

Transforming our Approach to Waste Heat

LoT-NET Workshop

Ammonia-Salt Resorption for
Thermal Transformations

13th December 2019

ERA ENERGY
RESEARCH
ACCELERATOR

EPSRC
Engineering and Physical Sciences
Research Council

TNO

WARWICK
THE UNIVERSITY OF WARWICK

Overview

- Problem statement
- Resorption and Thermal Transformers
- Experiments and Tests to date
- What is next?





Government reports identify
48TWh/yr of waste heat sources
from industry
Equivalent to a 1/6th of industrial
energy use

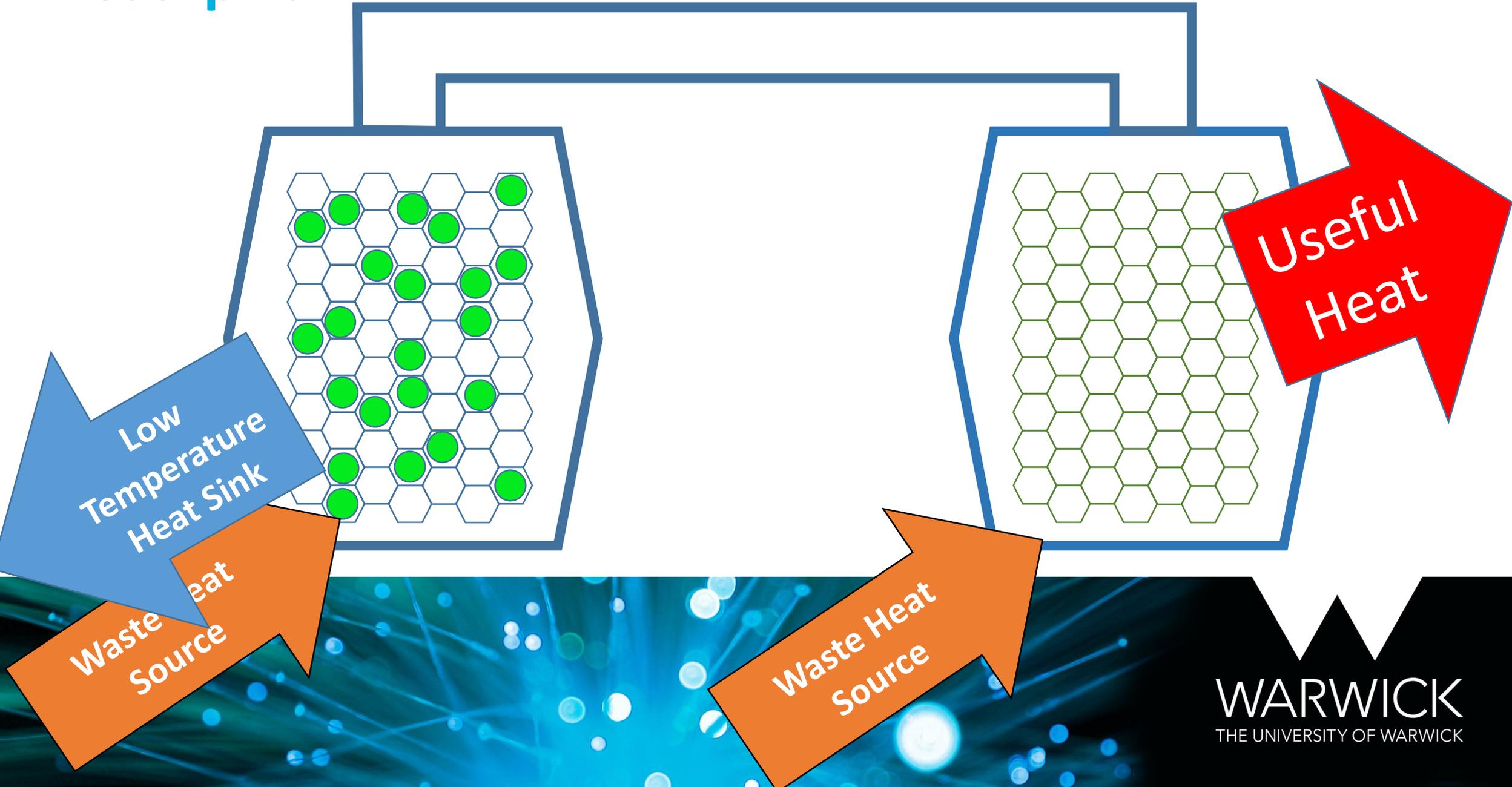
Element Energy (2014) Imperial College London for DECC

Resorption Thermal Transformers

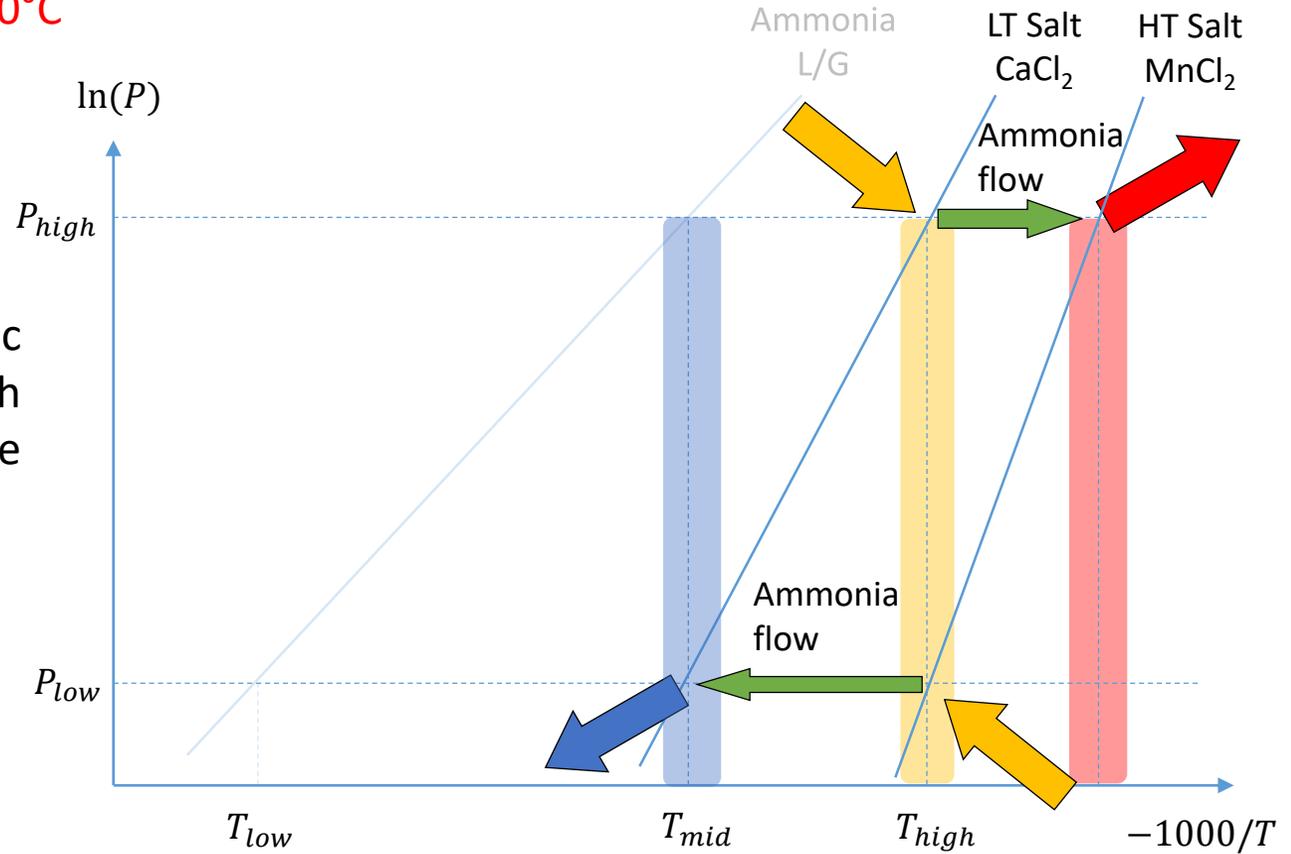
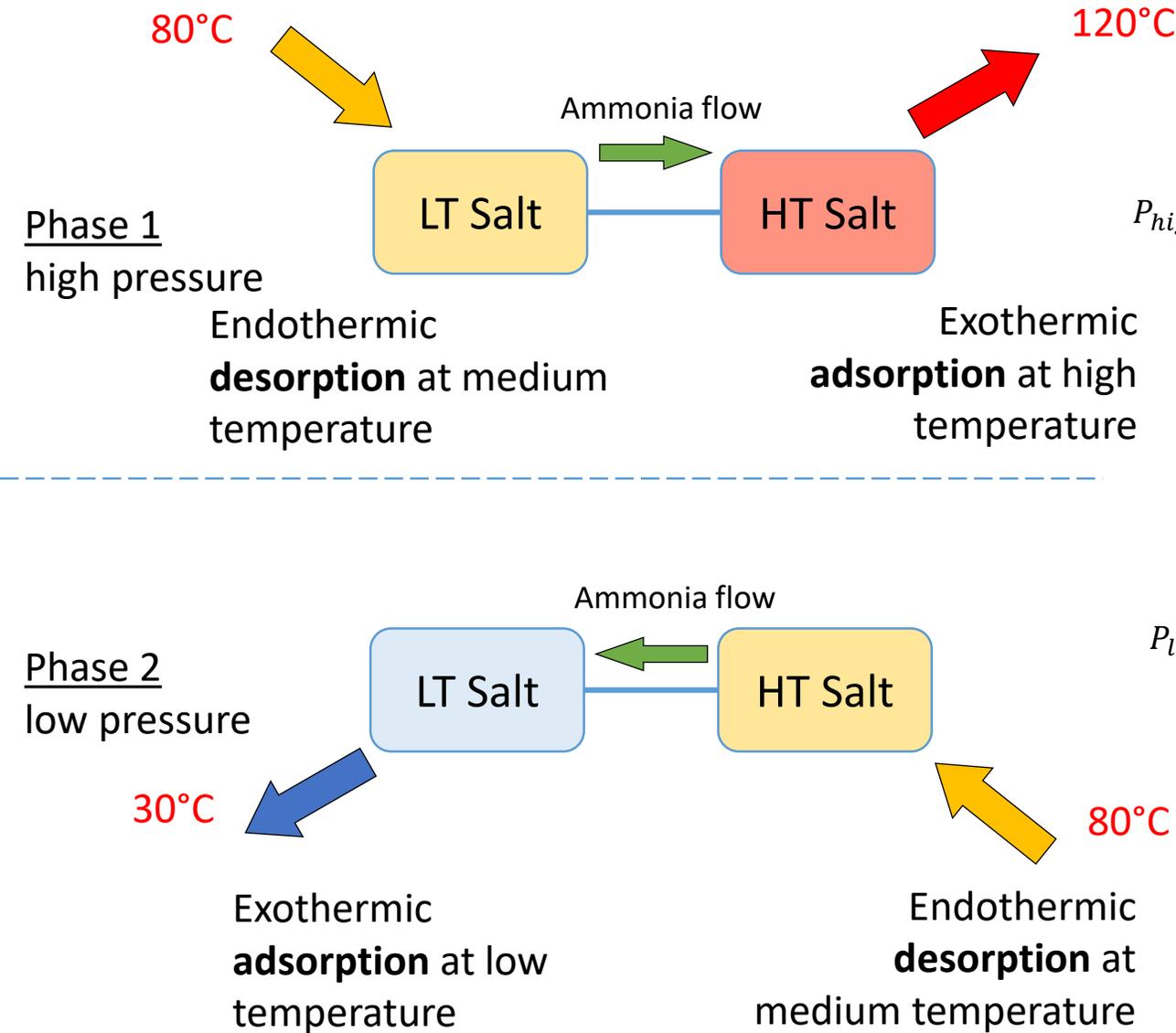
- Recovers waste heat by upgrading to useful temperature
- Simple concept and design, no pump, no evaporator or condenser
- Components can be cheap to manufacture
- Use of solid salts enables an endless list of possible operations and alternative applications



Resorption



2-Salt resorption cycle thermal transformer

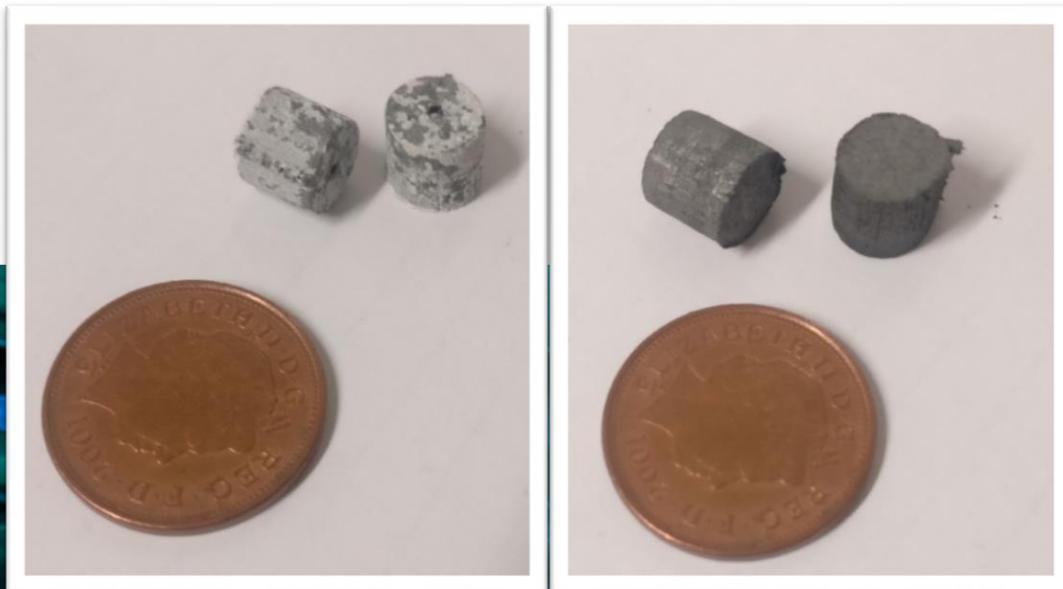


Route to thesis

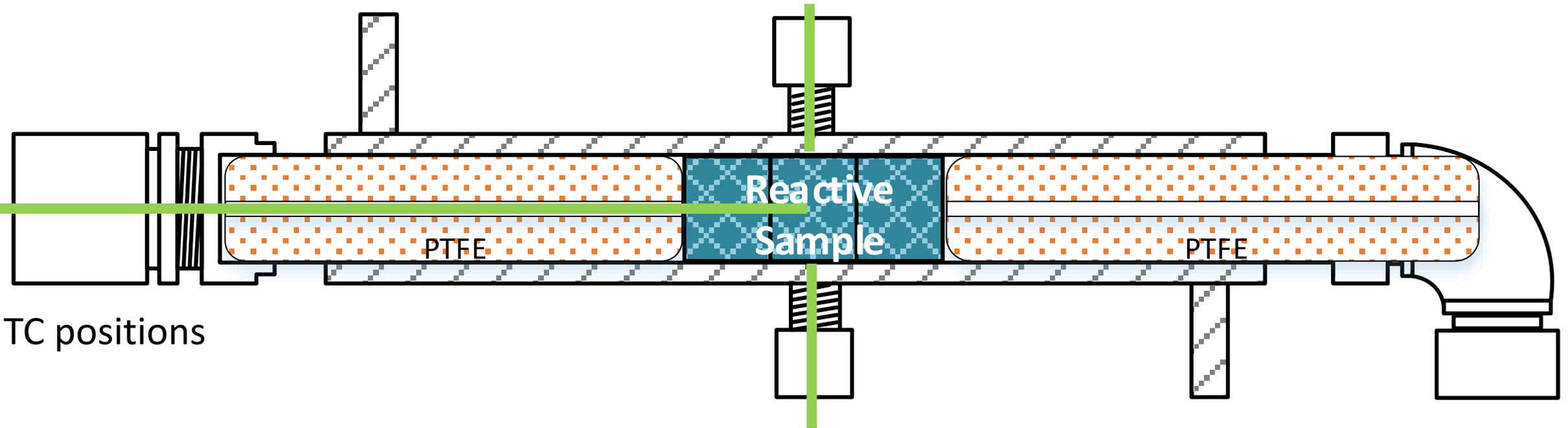
- Samples implanted in a conductive matrix of ENG
- Large Temperature Jump tests, to test the material under real working conditions
- Model the composite material behaviour
- Design a working transformer

Main tested samples with a content:

- 0.317kg salt/kg composite
- 0.531kg salt/kg composite



Large Temperature Jump Reactor



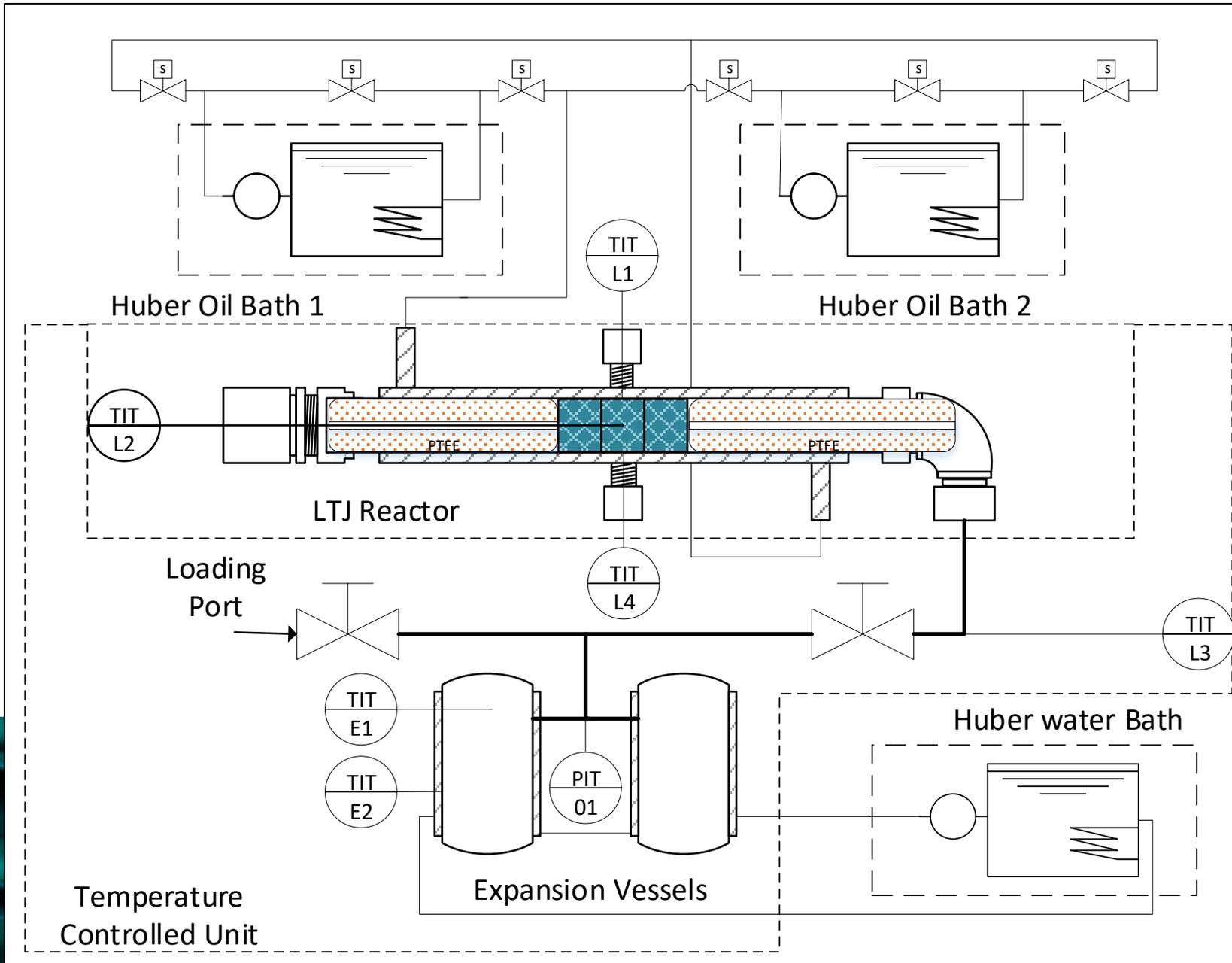
½ inch pipe
centre

1 inch pipe
jacket

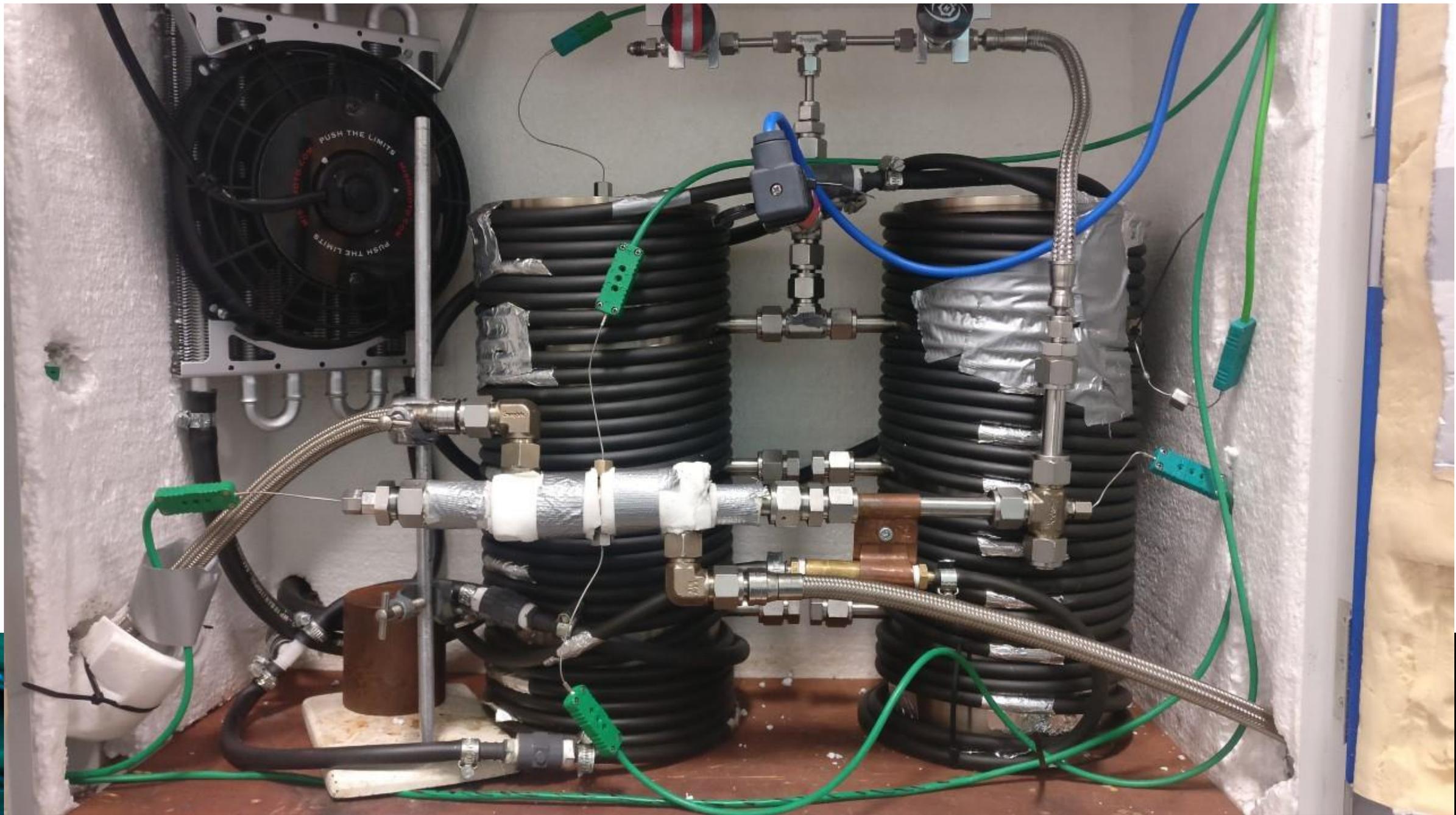
PTFE to fill
heated volume

O-ring Swagelok
face seal fitting

Large Temperature Jump Reactor

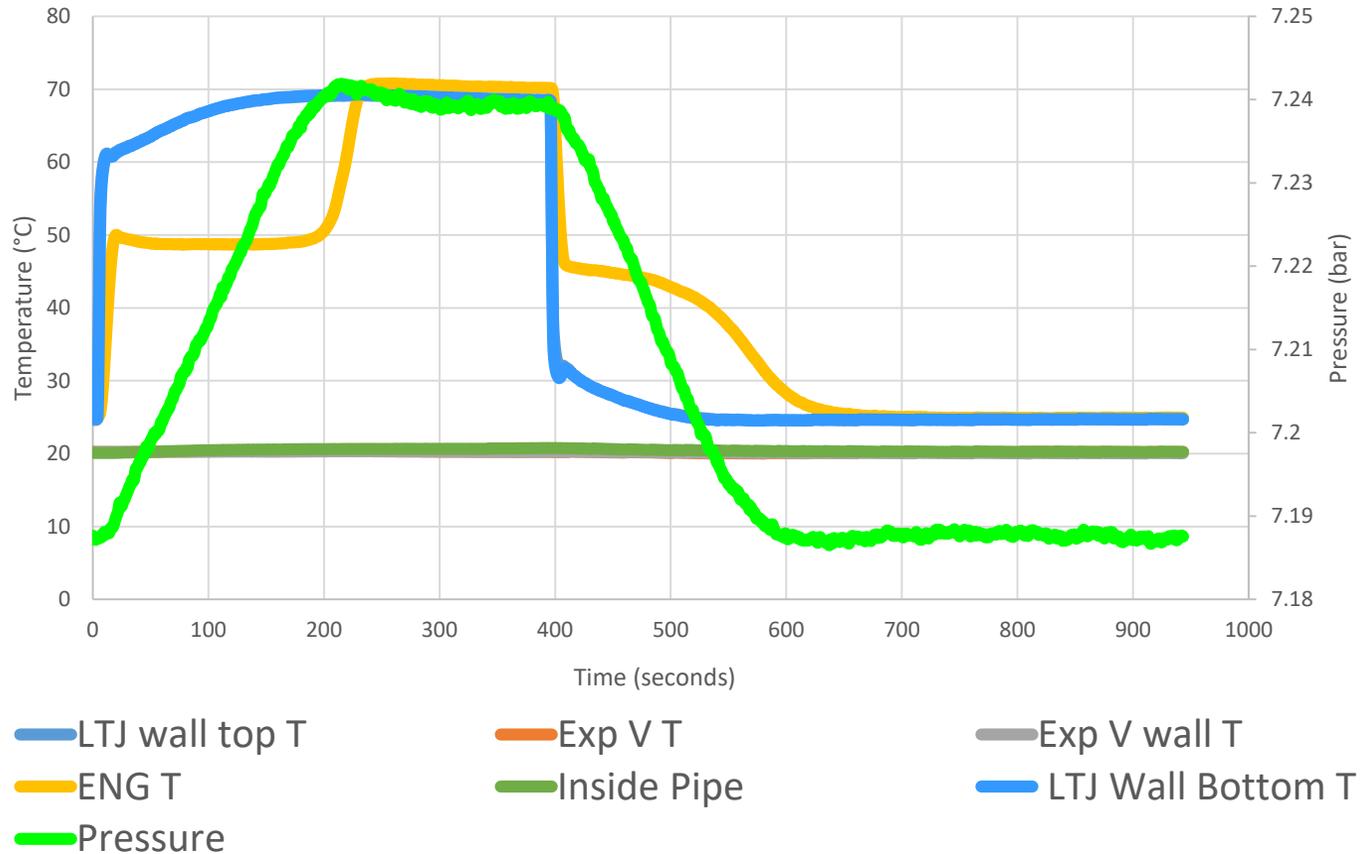


- ~12litres expansion vessels
- Huber baths with silicone oil
- Operate from -5 to 170°C
- Water bath controls expansion vessels and unit temperature



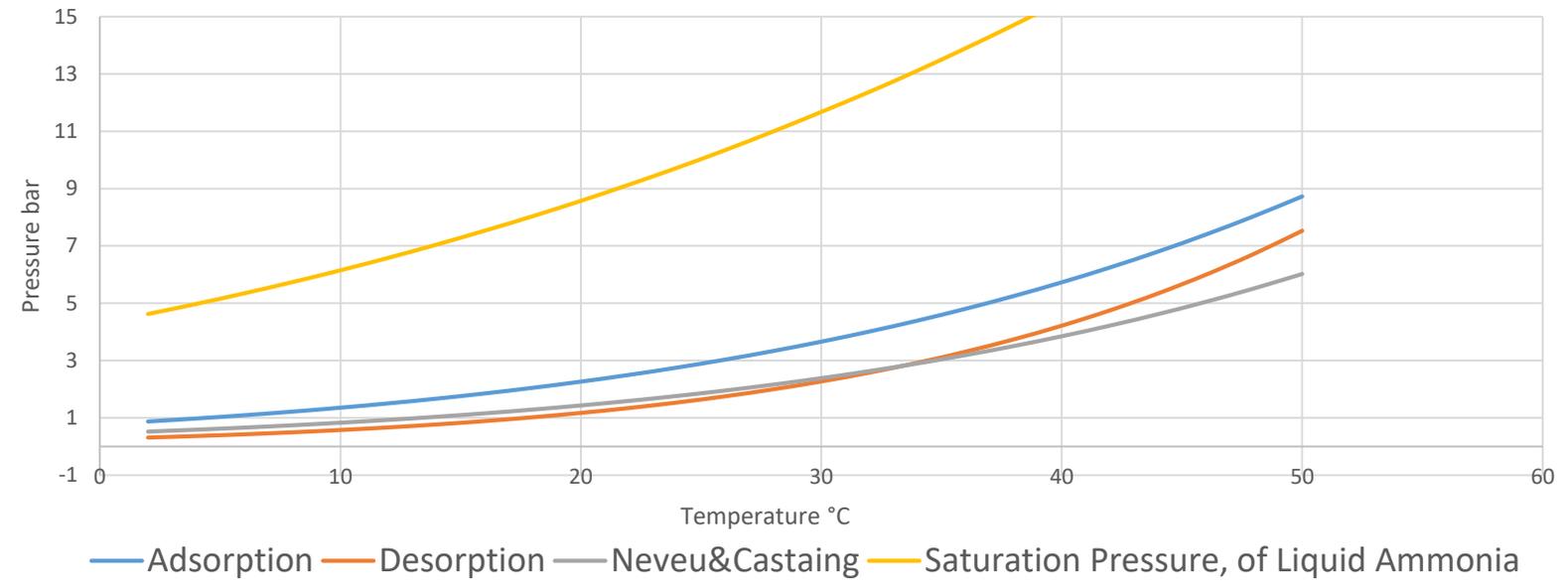
Results, full LTJ cycle

LTJ Experimental Cycle Result

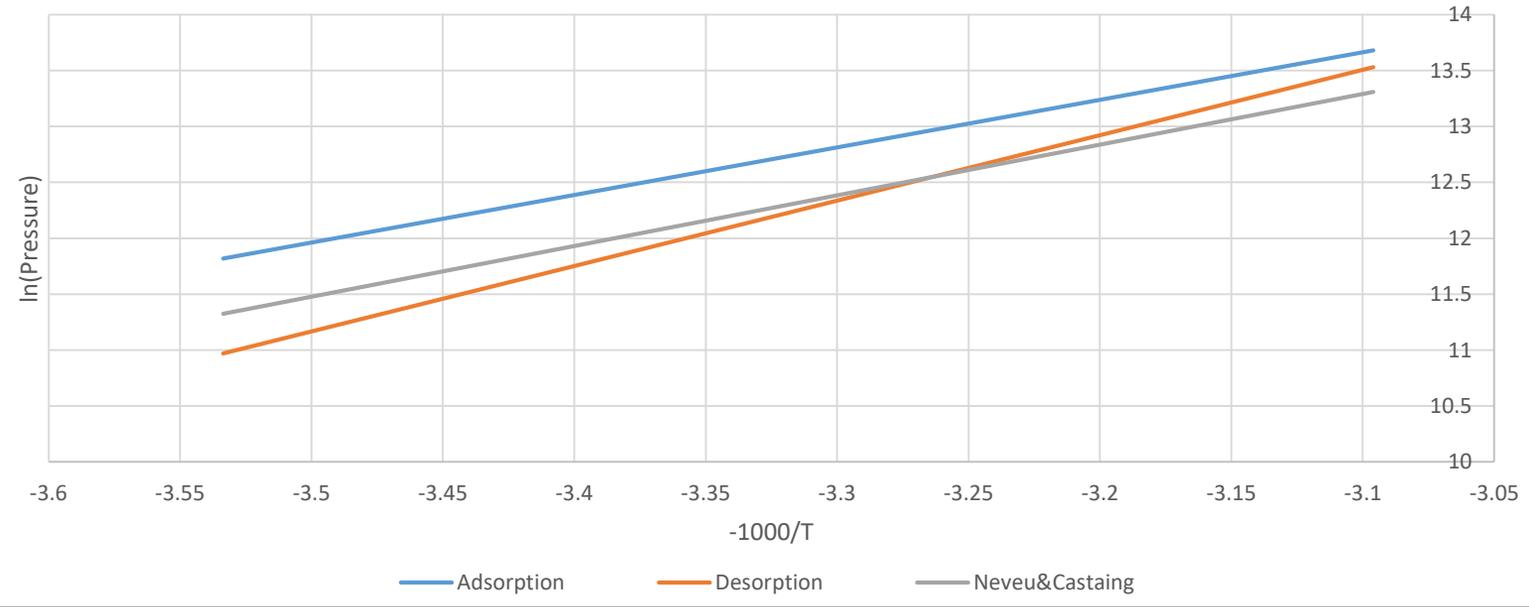


- Two scales on graph
- Repeated for different cases
- A new equilibrium line has to be calculated first based on position of phase change

Equilibrium Data



Comparison of Equilibria Data



Empirical reaction mode

$$\frac{dx}{dt} = [1 - x]^{y_0} \cdot Ar \cdot \frac{P - P_{eq}(T)}{P}$$

Order of reaction $y_0 = 2$

Arrhenius term $Ar = 3.5$

Pressure

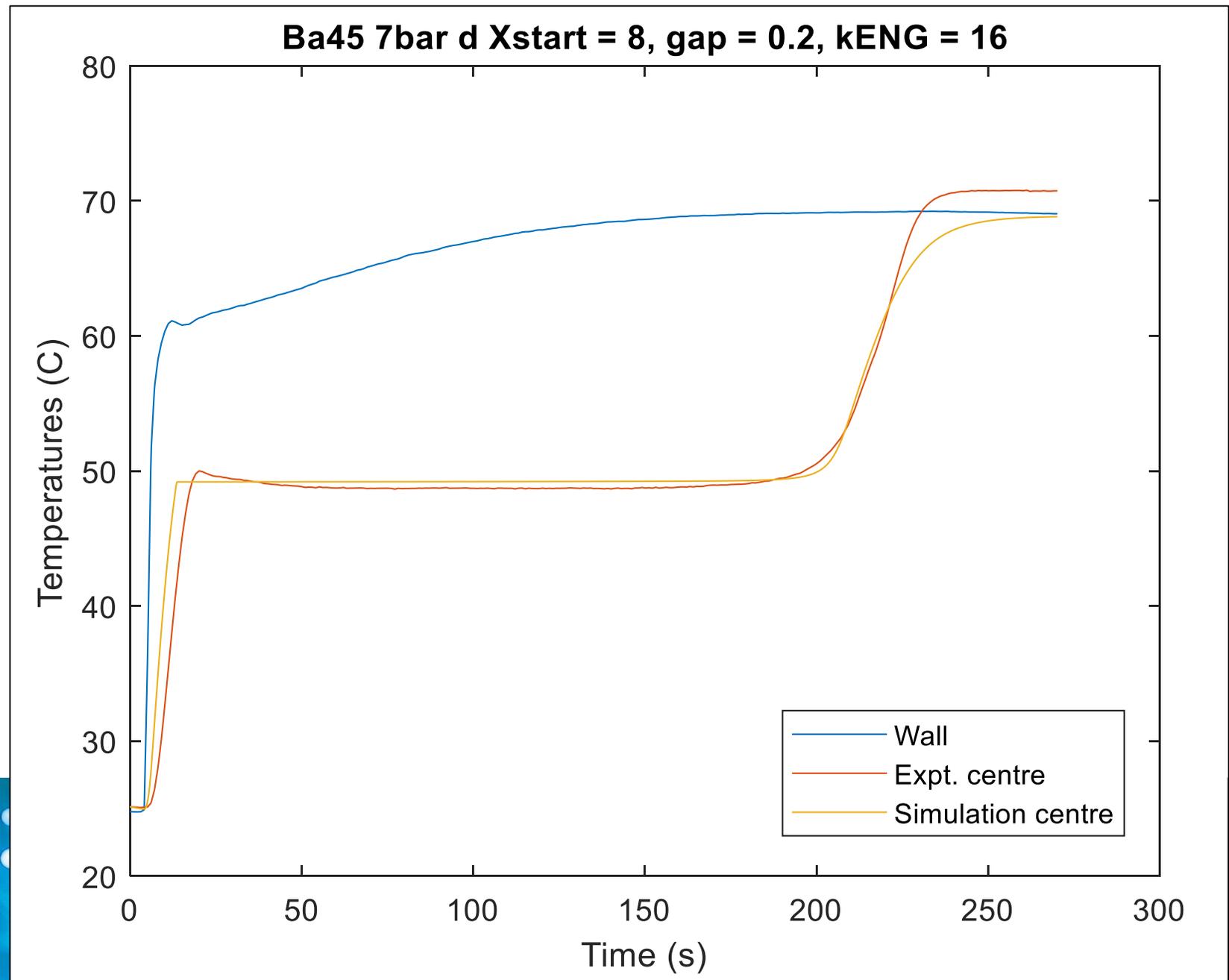
7bar

Mass Salt

0.759g

Mass Fraction of Salt

0.53g/g



Model Results

Order of reaction $y_0 = 1$
Arrhenius term $Ar = 0.1$

Pressure

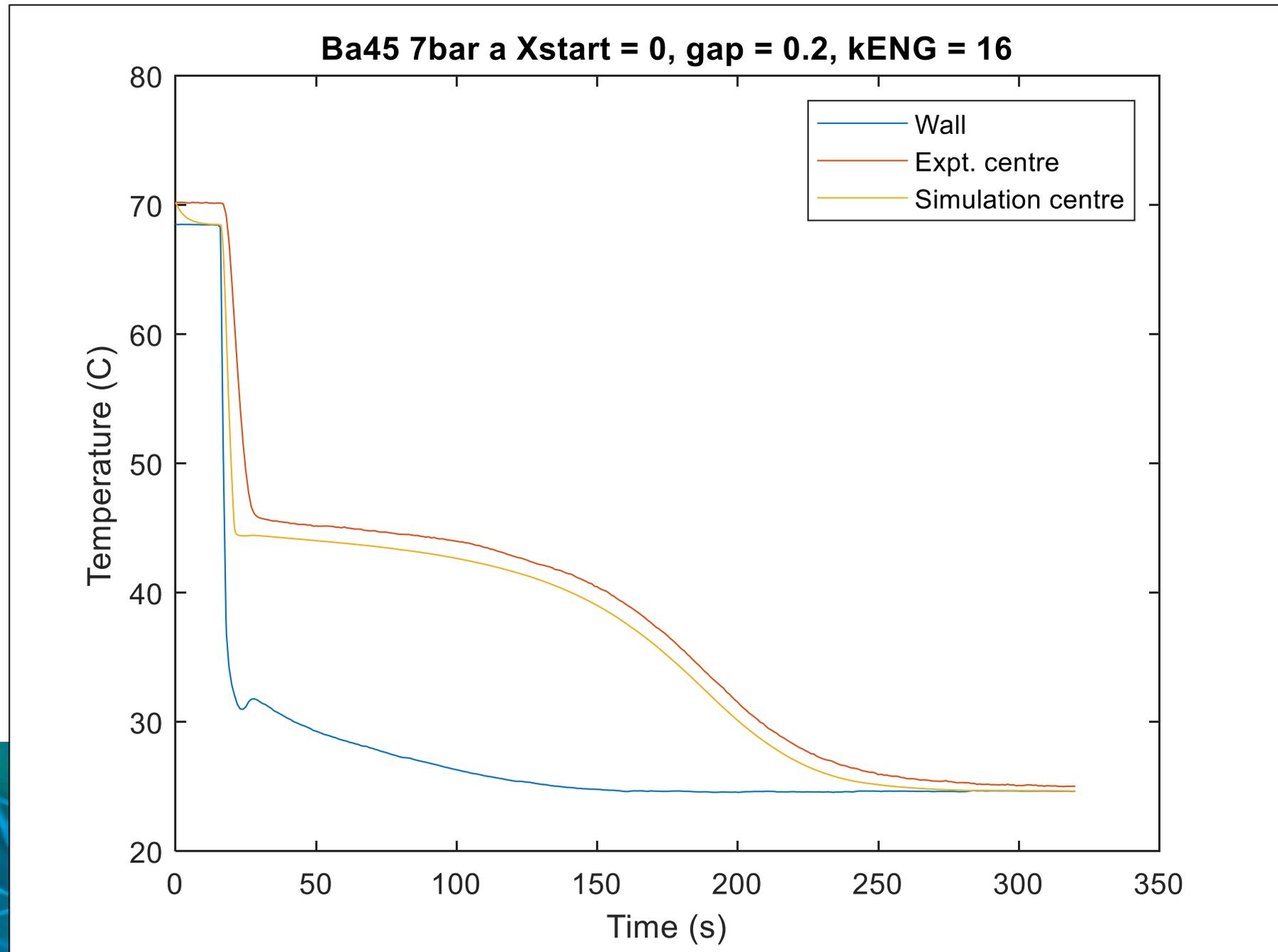
7bar

Mass Salt

0.759g

Mass Fraction of Salt

0.53g/g



To conclude..

- Salts and resorption enable effective utilisation of heat
- LTJ testing shows materials behaviour under real working conditions
- Modelling tests gives reaction data that enables development of working machines

Questions?

